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APPLICATION NO	FILING DATE	TIRSUNAMED INVENTOR	ALTORNEY DOCKEL NO	CONFIRMATION NO
09.938,277	08/23/2001	Barbara J. Hughey	NSL6382	1932
7;	90 02 13 2003			
Samuels, Gauthier & Stevens LLP Suite 3300 225 Franklin Street			EXAMINER	
			HASHMI, ZIA R	
Boston, MA 02110			ART UNIT	PAPER NUMBER
			2881	
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Please find below and/or attached an Office communication concerning this application or proceeding.

S Patent and Trademark Office PTO-326 (Rev. 04-01)

2) Notice of Draftsperson's Patent Drawing Review (PTO-948)

3) Information Disclosure Statement(s) (PTO-1449) Paper No(s) 4 & 6

Other:

-Application/Control Number: 09/938.277

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#### **DETAILED ACTION**

1. According to "Preliminary Amendment" received August 20, 2002, the first paragraph of the text has been replaced by a new one, as indicated.

## Claim Rejections - 35 USC § 102

2. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States
- 3. Independent claims 1, 21, 36, and 40 and dependent claims 2-5, 22-23, and 25 are rejected under 102(b) as being anticipated by Koudijs et al. (5,438,194).
- 4. With respect to claims 1 and 21 dependent claims 2-5, 22-23, and 25. Koudijs et al. disclose a method and apparatus of converting a non-gaseous sample for accelerator mass spectrometry analysis, comprising: converting desired elements present in the non-gaseous sample to a predetermined gaseous form ( col. 3. lines 63-67 and Fig. 1 and 2 ), and transporting the predetermined gaseous form to an accelerator mass spectrometer ion source ( col. 2. lines 12-14, col. 3. lines 40-43. Fig. 1 and 3 ). They also disclose that the step of converting comprises oxidizing the non-gaseous sample ( col. 4, lines 6-11 and 27-40 ), and that the step of chemically reacting comprises oxidizing and converting carbon in the sample to carbon dioxide ( col. 4, lines 34-40 ) by pyrolysis ( col. 3. lines 30-32 and col. 5, lines 11-13 ). Koudijs et al. further disclose a method of converting a non-gaseous sample for analytical processing.

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comprising: injecting the sample directly into a converter ( col. 3, line 52 and Fig. 2 ). which comprises of a pyrolizer ( col. 3, lines 31-32 and col. 5, line 13 ), converting the desired elements present in the sample to a predetermined gaseous form; and providing the predetermined gaseous form to an analytical device for processing ( col. 4, lines 4-11 ), comprising of an accelerator mass spectrometer ( col. 6, lines 11-12 ). They also disclose an interface for introducing non-gaseous sample as a predetermined gaseous form into an accelerator mass spectrometer, the interface comprising a first stage that receives the non-gaseous sample and separates an analyte from carrier material of the sample, to provide a separated sample stream that comprises the analyte ( 3 and 5 in Fig. 1 ); and a second stage that receives the separated sample stream converted into a gaseous form , and transports the gaseous form along a flow line to the accelerator mass spectrometer ( col. 3, lines 27-37, 3 and 5 in Fig. 1 ).

# Claim Rejections - 35 USC § 103

- 5. The following is a quotation of 35 U S C 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

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- 6. Claims 6-9, 10-20, 24, 26-35, 37-39, and 41-44 are rejected as being unpatentable over Koudijs et al. (5,438,194), in view of Zare et al. (4,988,879) and Hofstadler et al. (6,342,393).
- 7. With respect to dependent claims 7-9 and 13, Koudijs et al. fail to disclose method of converting desired elements present in non-gaseous form by depositing non-gaseous sample on a solid substrate, irradiating with a laser beam and desorbing the non-gaseous sample. Zare et al., however, disclose a method wherein prior to step of converting, the method comprises depositing the non-gaseous sample on a solid substrate ( col. 5, lines 41-42), and desorbing the non-gaseous sample from the substrate ( col. 5, lines 12-15, 43-49, col. 8, lines 20-24, col. 10, lines 1-5, and 102 in Fig. 1 ), by irradiating the sample with a laser beam ( col. 5, lines 45-49, col. 7, lines 17-20 and Fig. 1 ). In their method, volatile components are removed from the sample subsequent to the step of depositing, and prior to step of desorbing ( col. 14, lines 35-45. ).
- 8. With respect to claims 6, 10-12, 14-20, 24, 26-35, 37-39, and 41-44, Koudijs et al. fail to disclose method or apparatus in form of an interface for converting non-gaseous sample for analytical processing by nebulizing the sample using electrospray technique. However, Hofstadler et al. disclose a method of nebulizing a sample by using electrospary, in form of providing fine ionized spray to the converter, converting the desired elements present in the nebulized sample to predetermined gaseous form (col. 6, lines 57-60 and Fig. 1), and providing gaseous form to an analytical processing device, like a mass analyzer, for analysis (col. 6, lines 36-37, 42-43, and Fig. 1).

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It would have been obvious to one having ordinary skill in the art at the time of the invention was made to modify the method and apparatus of Koudijs. Zare and Hofstadler et al. and incorporate features like introduction of sub-micrometer sized particles to non-gaseous sample, or introducing sample into the converter using piezo-electric pipetter, because Hofstadler et al. teach (col. 1, lines 29-32) that the utility of mass spectrometer (MS) as an analytical tool is the availability of a variety of different MS methods, instruments and techniques, which can provide different pieces of information about the samples.

### Conclusion

- 6. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure. Meyer discloses (6.455,844) a method and apparatus for detecting single atoms of isotopes using accelerator mass spectrometry.
- 7. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Zia Hashmi whose telephone number is (703) 305-0419. The examiner can normally be reached between 8 30 AM- 5 PM. If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, John R. Lee can be reached on (703) 308-4116.

JOHN RALES
SUPERVISOR OF PETERS CHAMBLES

Zia Hashmi

January 14, 2002.